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balanced equa-tions apply to both chemistry and everyday
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288 seats, 864 wheels, 576 pedals 2. Answers will vary but
should include the correct number of “parts” to make the
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Chapter 11 Small-Scale Lab

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Chapter 11 Small-Scale Lab Section 11.3 Precipitation Reactions: Formation of Solids, page 345 Analysis 1. a. $\text{Na}_2\text{CO}_3 + 2\text{AgNO}_3$

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→ $2\text{NaNO}_3 + 3\text{Ag}_2\text{CO}_3(\text{s}) \rightarrow 6\text{NaNO}_3 + 3\text{Pb}(\text{PO}_4)_2(\text{s}) + 2\text{Ag}_2\text{O}(\text{s})$

12.1 The Arithmetic of Equations 12

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... excess of 1800 is a logical answer. • The unit of the known (FSW 3 HP 2) cancels. • The answer has the correct unit (W). 3 Evaluate Does the result make sense?

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Overview of Chemistry 1 Honors Chapter 12: Stoichiometry. Terms in this set (21) Stoichiometry. The calculation of quantities in chemical reactions is a subject of chemistry. Mole ratio. ... Use the following balanced equation to answer the question: $\text{Mg} + 2\text{H}_2\text{O} \rightarrow \text{Mg}(\text{OH})_2 + \text{H}_2$...

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